

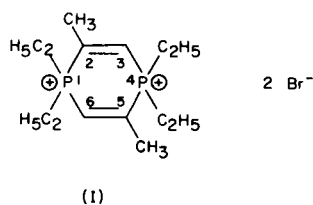
The Molecular structure of $(C_{14}H_{18}P_2)^{+2} 2Br^-$

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Sir:

The structure of the compound $(C_{14}H_{18}P_2)^{+2} 2Br^-$ has been shown to be 1,1,4,4-tetraethyl-2,5-dimethyl-1,4-diphosphoniacyclohexadiene-1,4-dibromide (I) by a single crystal x-ray diffraction study.



It has been predicted for the 1,1,2,4,4,5-hexaphenyl analogue of this compound (1,2) that delocalization of the π -electronic charge over the entire ring occurs. Support for this assumption is based upon ^{31}P nmr shifts which are positive (+3.5) indicating substantial shielding of the phosphorus nucleus whereas most simple phosphonium salts display ^{31}P shifts in the -20 to -30 cps region. (The ^{31}P nmr spectra are at 40.5 Mc/sec with the shifts in parts per million of methanol solution relative to an external standard of 85% phosphoric acid). Our x-ray data reported here, shows that this delocalization over the entire ring does not occur in the alkylated system (I).

This compound crystallizes in the space group $P2_1/c$ with two molecules in a unit cell of dimensions, $a = 7.73$, $b = 14.76$, $c = 10.55 \pm 0.02 \text{ \AA}$ and $\beta = 124^\circ 55' \pm 5'$. Refinement of the data from a single crystal of admittedly

poor quality has reached a value of $R = 0.16$. A difference Fourier map at this stage shows no peaks greater than 1.5 e/\AA^3 above background. The molecule has a center of symmetry and the six-membered ring is found to be planar within the estimated standard deviations of the atomic positions. The C_2-C_3 and C_5-C_6 distances are $1.30 \pm 0.5 \text{ \AA}$ proving that the ring is a diene. Phosphorus carbon distances within the ring are $1.82 \pm 0.5 \text{ \AA}$ whereas the phosphorus carbon distances outside the ring are $1.88 \pm 0.05 \text{ \AA}$. All other carbon-carbon distances average to $1.55 \pm 0.03 \text{ \AA}$. Bond angles throughout the structure conform to the expected values with the tetrahedral angles averaging $110 \pm 2^\circ$ and the angles around the sp^2 hybridized atoms averaging $120 \pm 4^\circ$. Additional crystals of the compound, suitable for diffractometer studies, are now under preparation.

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